



Virginia Department of Health

Bureau of Toxic Substances



P.O. Box 2448, 1500 East Main Street, Room 124
Richmond, Virginia 23218 (804)786-1763

FACT SHEET ON LITHIUM

GENERAL INFORMATION

Lithium (Li) belongs to sodium and potassium group of elements and, accordingly, has similar physical, chemical, and some biological properties. Lithium is an element of the alkali-metal group with atomic number 3 and atomic weight 6.94. It is widely distributed throughout the world in a variety of minerals. Natural waters may contain significant concentration of lithium (11 ppm in sea water and 1 ppm in certain mineral waters). Lithium is commonly found in many plants and animals tissues. Daily human intake is about 2 milligrams (mg) from some food sources.

USES

Lithium and its compounds are widely used in a variety of industries including metalurgy, ceramics, air conditioning, chemical and pharmaceutical manufacture, and for lubrication grease.

HEALTH EFFECTS

There is ample evidence that lithium at low levels causes no serious adverse health effects. Because lithium is routinely used as a drug for the treatment of clinical depression, there is a large body of data on the human health effects of lithium exposure. Lithium is readily absorbed from the gastrointestinal tract. Distribution in the human organs is almost uniform. Excretion is chiefly through the kidneys, but some is eliminated in the feces. In general the body distribution of lithium is quite similar to that of sodium, and potassium and it may be competing with sodium and potassium at the renal tubular level.

Lithium is administered therapeutically as the carbonate salt (Lithium carbonate) in daily oral doses of 900-1800 mg/day for the treatment of manic and endogenous depression. The optimum maintenance dosage is often determined by monitoring the serum concentration. Good correlation exists between the symptomology of lithium poisoning and the serum lithium concentration. No toxic effects occur at serum lithium level below 1.5 milliequivalent/l (MEQ/L). Mild toxicity can occur at levels from 1.5 to 2.5 MEQ/L. Levels in excess of 3.5 MEQ/l are potentially fatal. The therapeutic use of lithium carbonate may produce unusual toxic responses. These include neuromuscular changes (tremor, muscle hyperirritability, and ataxia), central nervous system changes (blackout spells, epileptic seizures, slurred speech, coma, psychosomatic retardation, and increased thirst), cardiovascular changes (cardiac arrhythmia, hypertension, and circulatory collapse), gastrointestinal changes (anorexia, nausea, and vomiting), and renal damage (albuminuria and glycosuria). These changes appear to be more frequent when the serum levels increase above 1.5 milliequivalent/liter (MEQ/l).

Lithium treatment is not recommended to patients with significant renal or cardiovascular disease, severe debilitation or dehydration, or sodium depletion and to patients receiving other medications (eg. diuretics), because of the risk of lithium toxicity is high in such patients. Lithium carbonate may cause fetal harm

